analysis showed it to be a mixture of three sterols. The mixture was analyzed using an instrument' fitted with a 0.63-cm. (0.25-in.) o.d.  $\times$  1.8-m. (6-ft.) glass column packed with Gas Chrom Q, 100-120 mesh, and coated with 5% OV-101. Helium was used (80 ml./min.) as the carrier gas, and the column was maintained at 250°. Cholestane was used as the internal standard. The sterols were identified as campesterol (15.6%), stigmasterol (6.2%), and  $\beta$ -sitosterol (78.2%) by comparison of the relative retention times of the eluted sterols with reference samples of authentic sterols.

#### REFERENCES

(1) M. Koczwara, Publ. Pharm. Commun., Pol. Acad. Sci., 1, 65(1949).

(2) C. Provost, Bull. Soc. Hist. Natur. Doubs., 62(3), 67(1959-

(3) S. P. Hiremath, V. S. Hosurkan, M. K. Razdem, and V. G. Mathad, J. Karnatak Univ., 14, 49(1969); through Chem. Abstr., 74, 79532j(1971).

(4) E. W. Stoller and E. L. Webber, *J. Agr. Food Chem.*, **18**, 361 (1970).

(5) O. T. Rotini and I. F. Navari, Agrochimica, 24, 277(1970);

through Chem. Abstr., 73, 106265n(1971).

(6) A. L. Bakke and N. D. Render, Amer. J. Bot., 29, 353(1942).
(7) E. A. Hedgeson and R. Konzak, N. Dak. Agr. Exp. Sta. Bull., 12, 71(1950); through Chem. Abstr., 45, 2132e(1951).

(8) V. E. Erspamer, Riv. Ital. Essenze Profumi Piante Offic., Olii Veg., Saponi, 28, 105(1946); through Chem. Abstr., 41, 2959 (1947).

(9) M. N. Voronina, Farmakol. i Toksikol., 29(1), 70(1966); through Chem. Abstr., 64, 14811d(1966).

(10) G. Eglinton, A. G. Gonzalez, R. J. Hamilton, and R. A. Raphael, *Phytochemistry*, 1, 89(1962).

(11) P. Boiteau, B. Pasich, and A. R. Ratsimamanga, "Les Triterpenoides en Vegetale et Animale," Gauthier-Villars, 1964, p. 125.

(12) H. Budzikiewicz, C. Djerassi, and D. H. Williams, "Structure Elucidation of Natural Products by Mass Spectrometry," vol. II, Holden-Day, San Francisco, Calif., 1964, p. 122.

## ACKNOWLEDGMENTS AND ADDRESSES

Received August 29, 1972, from the Department of Pharmacognosy and Pharmacology, College of Pharmacy, University of Illinois at the Medical Center, Chicago, IL 60612

Accepted for publication October 27, 1972.

A To whom inquiries should be directed.

# Synthesis of Potential Antineoplastic Agents XXII: Compounds Related to 1-Nitro-3-[(p-phenylbenzylidene)amino]guanidine

FRANK D. POPP

Abstract 
The nitroguanylhydrazone of 4-biphenylcarboxaldehyde exhibits antineoplastic activity in the Walker 256 and KB test systems. Other nitroguanylhydrazones and other derivatives of 4-biphenylcarboxaldehyde were prepared and found to be devoid of antineoplastic activity.

In connection with other work in progress in this laboratory, the nitroguanylhydrazone of 4-biphenyl-carboxaldehyde (I) was prepared. In routine screening, this compound was found to possess antineoplastic activity against Walker carcinosarcoma 256 (Table I) and had confirmed activity against KB cell culture. To explore this lead further, a series of nitroguanyl-

$$C_0H_3$$
 —  $CH$  —  $NH$  —  $C$  —  $NHNO_2$ 

hydrazones and some derivatives of 4-biphenylcarboxaldehyde were prepared for screening. Nitroguanylhydrazones were previously used to identify aldehydes and

Table I—Summary of Screening of 1-Nitro-3-[(p-phenylbenzylidene)amino]guanidine against Walker Carcinosarcoma 256 (Subcutaneous) in Fischer 344 Rats<sup>a</sup>

Vehicle	Day of First Injec-	ber of Injec-	Dose,	Animal —Perce Dif- ference T-C	
venicie	tion	tions	mg./kg.	1-0	1/0
Hydroxypropyl-	1	9	400	-6	150
cellulose	ī	ģ	200	-2	138
001101000	ī	9	100	$-\overline{2}$	116
	3	4	400	-1	166
	3	4	200⁴	1	177
	3	4	100	-2	116
Saline with	1	9	400	<b>-7</b>	144
polysorbate	1	9	200	-2	133
80	1	9	100	5	144
	1	9	<i>5</i> 0	1	150
	3	4	400	-8	144
	3 3 3	4	200	-4	127
	3	4	100	1	111

<sup>&</sup>lt;sup>a</sup> Supplied by Drug Research and Development, Chemotherapy, National Cancer Institute. Intraperitoneal administration was made daily, with evaluation on Day 30. In all cases, there was 6/6 survivors. <sup>b</sup> Average weight change of test group minus average weight change of control animals in grams. <sup>c</sup> Ratio of survival time of treated to control animals expressed as percent. <sup>d</sup> Klucel. <sup>e</sup> Two cures.

<sup>4</sup> Perkin-Elmer model 881.

Table II--Nitroguanylhydrazones

R C=N-NH-C-NHNO<sub>1</sub>

Aldehyde or Ketone Used (RR <sub>1</sub> CO)			Analysis, %		
	Melting Point	%	Formula	N Calc.	N Found
4-Biphenylcarboxaldehyde	225-227°	90	C14H12N6O2	24.72	24.40
Acetylferrocene	205-207°	80	C18H15FeN5O2	21.28	20.95
Acetylisatin	224226°	30	C <sub>11</sub> H <sub>10</sub> N <sub>2</sub> O <sub>3</sub> a	28.96	28.93
1-Benzylindole-3-carboxaldehyde	208~209°	91	$C_{17}H_{16}N_{6}O_{2}$	24.99	24.76
4-Bromobenzaldehyde	220-221°	92	CaHaBr NaOa	24.48	24.41
Cyclopentanone	13 <b>9</b> 141°	65	$C_6H_{11}N_5O_2$	37.82	37.74
Ferrocenecarboxaldehyde	195-196°	76	C12H13FeN5O2	22.23	22.05
4-Fluorobenzaldehyde	206–207°	83	C <sub>8</sub> H <sub>8</sub> FN <sub>5</sub> O <sub>2</sub>	31.10	30.94
Isatin	268-270°	98	C <sub>9</sub> H <sub>8</sub> N <sub>6</sub> O <sub>3</sub> b	33.86	33.87
Pentafluorobenzaldehyde	233–234°	94	CaH4F5N5O2	23.57	23.41
Pyridine-3-carboxaldehyde	230~231°	67	C7H8N6O2	40.37	40.50
Quinoline-2-carboxaldehyde	228-2 <b>29</b> °	61	C11H10N6O2	32.55	32.40
3,4,5-Trimethoxybenzaldehyde	208-209°	76	$C_{11}H_{15}N_{5}O_{5}$	23.56	23.46

Calc.: C, 45.52; H, 3.47. Found: C, 45.36; H, 3.61. Calc.: C, 43.55; H, 3.25. Found: C, 43.68; H, 3.39.

Table III-4-Biphenylcarboxaldehyde Derivatives

R-N=CH-	$\langle$	D)	≻—C <sub>0</sub> H <sub>1</sub>
---------	-----------	----	---------------------------------

-	Yield,			Analysis, %	
RNH <sub>2</sub> Used	Melting Point	%	Formula	N Calc.	N Found
Diaminomaleonitrile	231-232°	83	C <sub>17</sub> H <sub>12</sub> N <sub>4</sub> a	20.58	20.36
1,1-Dimethylhydrazine	86-87°	63	C15H15N2	12.49	12.41
4-Phenylsemicarbazide	207-208°	95	$C_{20}H_{17}N_{2}O$	13.32	13.40
4-Phenyl-3-thiosemicarbazide	197198°	99	C20H17N3S	12.68	12.66

a Calc.: C, 74.98; H, 4.48. Found: C, 74.81; H, 4.35.

ketones (1) and showed antitubercular activity (2). Previously unreported compounds in these two series are shown in Tables II and III.

Screening results indicate a complete lack of antineoplastic activity (T/C of 125% for tumor survival systems and T/C 42% for tumor weight-inhibition systems) for all except the title compound in Tables II and III. Furthermore, of a large number of known nitroguanylhydrazones that have been screened, only three showed even marginal activity. KB cell culture activity was observed, however, in some of these compounds.

#### EXPERIMENTAL<sup>1</sup>

Nitroguanylhydrazones-To a warm solution of 1.19 g. (0.01 mole) of 1-amino-3-nitroguanidine in 10 ml. of ethanol, 10 ml. of water, and 20 ml. of acetic acid was added 0.01 mole of the carbonyl compound in 25 ml, of ethanol. The mixture was heated on

the steam bath for 30 min., cooled, and filtered to give, after recrystallization from ethanol, the compounds listed in Table II.

4-Biphenylcarboxaldehyde Derivatives—To 1.82 g. (0.01 mole) of 4-biphenylcarboxaldehyde in 20 ml. of ethanol was added 0.01 mole of the appropriate carbonyl reagent in ethanol. The mixture was heated on the steam bath for 30 min., cooled, and filtered to give, after recrystallization from ethanol, the compounds listed in Table III.

### REFERENCES

(1) W. F. Whitmore, A. J. Revukas, and G. B. L. Smith, J. Amer. Chem. Soc., 57, 706(1935).

(2) W. D. Kumler and P. P. T. Sah, J. Amer. Pharm, Ass., Sci. Ed., 41, 375(1952).

## ACKNOWLEDGMENTS AND ADDRESSES

Received September 13, 1972, from the Department of Chemistry, Clarkson College of Technology, Potsdam, NY 13676

Accepted for publication October 16, 1972.

Supported in part by Grant IC-24 from the American Cancer Society.

Screening results were supplied by Drug Research and Development, Chemotherapy, National Cancer Institute, and the assistance of Dr. Harry B. Wood in this regard is acknowledged.

<sup>&</sup>lt;sup>1</sup> Analyses were performed by Spang Microanalytical Laboratory, Ann Arbor, Mich. Melting points were determined in capillary tubes and are corrected.